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Sodium sulfate is a typical electrostatically bonded ionic sulfate. The existence of free sulfate ions in solution is indicated by the easy formation of insoluble sulfates when these solutions are treated with Ba²⁺ or Pb²⁺ salts: Na₂SO₄ + BaCl₂ → 2 NaCl + BaSO₄. Sodium sulfate is unreactive toward most oxidizing or reducing agents.

Sodium sulfate - Wikipedia

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