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4 _____ . Ternary Ionic Compounds. • Made up of _____ elements. • Name the _____ then name the _____ without changing the ending to "ide". Examples: Na. 2SO.

4 _____ FeCrO.

4 _____ . CHEMISTRY: A Study of Matter.

Note Taking Guide: Episode 603 Name Binary Ionic Compounds

View full document. CHEMISTRY: A Study of Matter © 2004, GPB 6.1 Note Taking Guide: Episode 603 Name Binary Ionic Compounds Consist of only two _ elements. Name the positive ion (the _ cation). Name the negative ion (the _ anion), changing the ending to ide _ . When metals that can form more _ than one type of Ion are in a compound, use a Roman numeral _ in parentheses after the name of the metal to show the charge .

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