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The ideal gas model is used to predict changes in four related gas properties: volume, number of particles, temperature, and pressure. Volumes of gases are usually described in liters, L, or cubic meters, m<sup>3</sup>, and numbers of particles are usually described in moles, mol. Although gas temperatures are often measured with thermometers that

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2.2 If the gas's volume is changed to  $\frac{1}{2} V$  and its temperature to  $2T$ , what will the gas's pressure be? 2.3 If the gas's volume is changed to  $4 V$  and its pressure to  $\frac{1}{4} V$ , what will its

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temperature be? Question 3 A 2 cm<sup>3</sup> gas bubble rises from the bottom of the sea, where the pressure is 275 kPa and the temperature 5°C.

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