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Mastering Chemistry - Chapter 4. For soluble salts, the molarity of a particular ion is equal to the molarity of that compound times the subscript for that ion. For example, 1 M of AlCl3 is 1 M in Al³⁺ and 3 M in Cl⁻. 1 M of (NH4)2SO4 is 2 M in NH4⁺ and 1 M in SO4²⁻. Chapter 4 Mastering Chemistry Answer Key - examget.net

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