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Nomenclature changes and the adoption of IUPAC periodic table conventions are highlights of the narrative revisions, along with changes to the discussion of d orbitals. In-text examples have been reformatted to facilitate learning, and the accompanying Interactive Examples in OWLv2 have been redesigned to better parallel the problem-solving approach in the narrative. New Capstone Problems have been added to a number of chapters. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

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Chemistry.

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structure, molecular orbitals, hybridization of atomic orbitals in carbon, chemical equilibrium, enthalpy, and acids and bases. The chapters that follow focus on the compounds of carbon such as alkanes and cycloalkanes; benzene and other aromatic hydrocarbons; amines and other heterocyclic molecules; aldehydes and ketones; carboxylic acids and their derivatives; nucleic acids; amino acids; peptides; and proteins. The use of instrumentation methods in organic chemistry, particularly mass spectrometry and nuclear magnetic resonance spectroscopy, is also considered. An account of the mechanisms of an organic reaction is presented, paying particular attention to displacement and elimination reactions. This book concludes with a commentary on how most of the amino acids, sugars,

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heterocyclic molecules, and fatty acids necessary for life processes could have been formed on Earth. This book is intended for nonmajors taking an introductory organic chemistry course of two quarters or one semester in length.

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living systems and the physical world around us to illustrate crucial concepts.

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