

## Heat Calculations Answer Key If8767 Pg 25

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~~2) Solve for the heat required to change the water into steam (no change in temp). 3) Calculate the heat required to change the temperature of the steam from 100.0 oC to 110.0 oC. 4) To get the heat required for the whole process, \_\_\_\_ the calculated heats from above. Q = m x ΔH vap Q = 1000. g x 2260 J/g = 2,260,000 J Q = m x C x Δt~~

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~~13-06a,b,c Heat and Heat Calculations wkst-Key Created Date: 4/28/2016 8:10:49 AM Boyertown Area School District / BASD Homepage Specific Heat Calculations Worksheet. In a heat calculation problem, if the problem asks about melting/freezing you would multiply the mass times \_\_\_\_\_. heat of fusion. heat of vaporization. or specific heat.~~