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nitrogen gas, 21% oxygen gas, and 1% other gases by volume.

Atmospheric pressure decreases higher up in the atmosphere.

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Answer. Molar mass = 151.1 g/mol Potassium:  $(39.10 / 101.1) \times 100 = 38.67\%$  Nitrogen:  $(14.01 / 101.1) \times 100 = 13.86\%$  Oxygen:  $(48.00 / 101.1) \times 100 = 12.58\%$ . Molar mass = 164.1 g/mol Potassium:  $(39.10 / 101.1) \times 100 = 38.67\%$  Nitrogen:  $(14.01 / 101.1) \times 100 = 18.96\%$  Oxygen:  $(48.00 / 101.1) \times 100 = 47.48\%$ .

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Atmospheric pressure decreases higher up in the atmosphere.

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