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Bonding. If the statement is

true, write "true." If it is

false, change the underlined

word or words to make it

true. Write your answer on

the line provided. ____F,

covalent_ 1. A group of

atoms united by ionic bonds

is called a molecule.

____T____ 2. A covalent

bond is formed by a shared

pair of ...

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7-2 Review and Reinforcement (continued) Write Lewis structures for each of the following molecules.

Indicated the bonds with either dots or dashes. 15.

NH₃ 18. PCl₃. 16. H₂ 19.

CCl₄. 17. C₂H₄ 20. C₂H₆.

Answer the following questions as directed. 21.

Explain why the molecule SF₄ is an exception to the octet rule.\

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S sunlevel 5-2 Review and Reinforcement (continued)
Identify each of the following elements as a metal (M), nonmetal (NM), or semimetal (SM).
M 19. sodium
SM 20. silicon
NM 21. neon
M 22. calcium
NM 23. nitrogen
Write the family names that have been given to each of the following groups.

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**Supplemental Vocabulary listed in the

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