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2? A 2+ B 3+ C 2? D 3? 4.

Fe. 2. O. 3 + x. CO . y. Fe

+ 3CO. 2. This equation will

be balanced when. A . x = 1

and . y = 2 B . x = 2 and .

y = 2 C . x = 3 and . y = 2

D . x = 2 and . y = 3. 5. An

acidic solution contains A

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only hydrogen ions B only

hydroxide ions C more

hydrogen ions than hydroxide

ions D more hydroxide ions

than hydrogen ions. 6.

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Answer all questions. 1. The
table below shows elements,

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Atomic numbers and their melting points. The letters do not represent actual symbols of the elements. Study it and answer the questions.

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2 (g) + 3CO₂ (g) + 4H₂O (?)

Name the type of enthalpy change which the student measured. No marks should be awarded for 'burning' since the word 'burned' appears in the text. (o) Unless the question is clearly about a

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non-chemistry issue, eg
costs in industrial chemical
process, a non-chemical
answer gains no marks.

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Finalised Marking

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Marks: 70 Time: 3 hours The
figures in the margin

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Indicate full marks for the
questions 1...

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3. Write the IUPAC name of the compound [1] Answer: 3-aminebutan-1-al. Question 5.

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2 by electricity /
(electric) current / (flow
of) electrons Mark
independently b A = chlorine
/ Cl₂ Ignore Cl 3 B =
hydrogen / H₂ Ignore H C =
sodium hydroxide / NaOH

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Ignore references to sodium chloride If both name and formula given, both must be correct, but ignore Cl and H Award 1 mark for chlorine and hydrogen the wrong way round

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Filter. Done. Clear all

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